

Coal to Liquids (CTL)

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The rising costs of fuel, combined with the fact that the U.S. has larger reserves of coal than it does oil, have prompted an upsurge of interest (<http://www.worldcoal.org/coal/uses-of-coal/coal-to-liquids/>) in the possibility of generating liquid fuels (such as gasoline and diesel) from coal in a process referred to as “coal-to-liquids (CTL)”. Coal is heavily used for electricity generation, but transportation currently relies almost entirely on oil-based fuels.

Coal can either be directly liquefied through processes called hydrogenation or carbonization, or it can first be first turned into “syngas (<http://en.wikipedia.org/wiki/Syngas>)” and then re-liquefied into appropriate fuels. The largest commercial-scale CTL project in the world (by Sasol (<http://www.sasol.com/>) in South Africa), uses the syngas method. The only commercial-scale project in the US, the Dakota gasification plant (http://www.dakotagas.com/About_Us/index.html) in North Dakota, produces large amounts of natural gas from coal.

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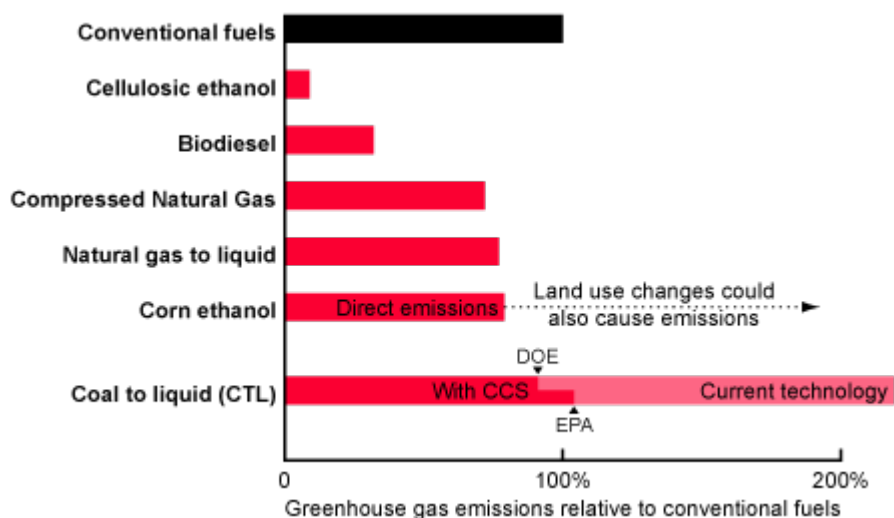
While CTL technology is not new, large scale application of the technology faces significant financial difficulties, and comes with significant environmental costs. CTL plants are very expensive to build, use a large amount of water to run, and release high levels of carbon dioxide (CO₂) into the atmosphere.

In addition to these direct concerns, an increase in the use of CTL fuels would fuel a vast increase in coal mining. It takes one ton (<http://www.netl.doe.gov/publications/others/pdf/Hirsch042506.pdf>) of coal to make two barrels of fuel. To replace 10% of US oil consumption with coal, using CTL, domestic coal mining would need to increase by 42% (<http://www.nrdc.org/energy/drivingithome/drivingithome.pdf>), an increase of 475 million tons per year. This dramatic boost in the scale of coal mining ([CoalMining.html](#)) would bring with it fish and wildlife habitat destruction, and increased emissions of mercury ([CoalMercury.html](#)), sulfur, methane, nitrous oxides, and CO₂, among other environmental and health concerns ([CoalTrueCost.html](#)).

CO₂ (carbon dioxide) emissions of CTL

Any use of coal, whether burning it for electricity, or creating liquid fuel, releases a number of pollutants into the atmosphere, including CO₂. These CO₂ emissions contribute to the problems of climate change, including increased storminess, droughts, and sea level rise. However, CO₂ emissions from a CTL plant would most likely far exceed the emissions generated by creating the same amount of fuel from oil through the traditional refining process (see figure).

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CLIMATE FOOTPRINT OF CTL (</figures/ClimateFootprintCTL/>) — What is the impact on the climate of different fuels?

It is theoretically possible to prevent these carbon dioxide emissions through carbon capture and sequestration (CCS) ([LowCarbonCoal.html](#)) technology. Manufacturing liquid fuels from coal allows for easier capture of carbon than simply burning coal, but this technology to capture CO₂ emissions is not yet commercially viable, and faces major cost and feasibility hurdles. It's uncertain whether CCS will ever be a viable technology, and if commercial-scale CCS does become available, it won't be for a long time.

Also, preventing CO₂ emissions from the fuel refining process does not address the problem of emissions from either generating the power to run the CTL plant or burning the resulting fuel. Therefore, the total CO₂ pollution that would be created in the production and burning of coal-derived liquid fuels is much higher than for oil-based fuels. The coal-based fuels would still be more polluting even if commercial carbon

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capture technology was developed and used in the CTL plants (based on a 2007 EPA estimate) (see figure). Additional emissions would also be produced in the course of mining the coal.

A 2009 report (<http://www.netl.doe.gov/File%20Library/Research/Energy%20Analysis/Coal/CBTL-Final-Report.pdf>) 4.1 Mb by the Department of Energy has updated some of these findings and suggests that CTL with CCS could produce fuels with slightly lower CO₂ emissions than oil-based fuels. This could only be true if the as-yet-undeveloped CCS technology proves to be commercially feasible and is implemented.

Water Consumption of CTL

The CTL refinery process consumes vast quantities of water to keep machinery from overheating, 5 to 7 gallons of water (http://www.sandia.gov/energy-water/congress_report.htm) for every gallon of fuel produced. For example, the recently cancelled (http://www.sourcewatch.org/index.php?title=Malmstrom_Air_Force_Base_Coal-to-Liquids) 22,000 barrel per day CTL plant at Malmstrom Air Force Base (<http://www.malmstrom.af.mil/>), Montana, would have consumed an estimated 1.7 billion gallons of water per year, enough to provide for 26,000 people. This scale of water use has potentially large impacts on a watershed, both through the depletion of existing rivers and through the outflow of large volumes of heated water.

Economics of CTL

A large (50,000 barrels per day) CTL plant is estimated to cost from \$3.6-6.0 billion (<http://files.asme.org/asmeorg/Communities/Technical/Energy/16088.pdf>). These high expenses make companies hesitant to invest in the technology unless they can get guaranteed contracts to buy the fuel they produce. The US Air Force has expressed a serious interest in CTL (http://online.wsj.com/article_email/SB121134017363909773-lMyQjAxMDI4MTIxMTMyNDEwWj.html) fuels, but is hampered by provision 526 in the 2007 Energy Bill (http://en.wikipedia.org/wiki/Energy_Independence_and_Security_Act_of_2007) that prohibits all federal agencies from using any alternative fuel with larger lifecycle greenhouse gas emissions than traditional fossil fuels.

In addition, because no CTL plant has ever used CCS, there is no precedent for estimating true costs. It can take 8-9 years to build a large-scale CTL facility, during which time price changes in raw materials and pending carbon tax legislation may significantly increase costs. With the high costs of a CCS-equipped CTL facility (not even accounting for a carbon tax (http://en.wikipedia.org/wiki/Carbon_tax)), these fuels would only be economically feasible if the price of crude oil rises to more than \$86 a barrel (estimated Jan 2009). And even then, it would be feasible only if coal prices do not rise along with oil prices. However, during the 2008 oil price surge, the cost of most types of coal rose steeply (<http://www.eia.doe.gov/cneaf/coal/page/coalnews/coalmar.html>) as well.

CTL in Alaska

Because Alaska possesses large coal reserves and has relatively high fuel prices, there have been several recent proposals for CTL projects within the state. The most prominent of these are the proposed [Fairbanks CTL \(FairbanksCTL.html\)](#) project, the [Beluga CTL \(BelugaCTL.html\)](#) project, the [Healy CTL \(HealyCTL.html\)](#) project, and the [Tyonek CTL \(/Issues/AlaskaCoal/TyonekCTL.html\)](#) project. Furthermore, the [US Defense Logistics Agency \(http://www.dla.mil/default.aspx\)](#) is working on a pilot program ([http://homernews.com/stories/032509/business_bu_009.shtml](#)) to solicit other projects that could provide CTL fuels to military bases throughout the state, assuming that CTL was determined to not run afoul of the 2007 Energy Bill restriction. In Spring 2011 a Texas-based energy company expressed interest ([http://www.peninsulaclarion.com/stories/041511/new_815352633.shtml](#)) in developing a CTL project somewhere in Cook Inlet *[CO₂]: carbon dioxide

Further Reading

- > [Department of Energy: National Energy Transportation Laboratory. 2006. Economic Impacts of U.S. Liquid Fuel Mitigation Options. \(http://www.netl.doe.gov/publications/others/pdf/Hirsch042506.pdf\)](#)
- > [Cook Inletkeeper overview of CTL in Alaska \(http://inletkeeper.org/energy-and-alaska/coal/coal-to-liquids\)](#)

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> U.S. Department of Energy, Office of Fossil Fuel Energy, Office of Sequestration, Hydrogen, and Clean Fuels. 2008. Coal Conversion Technology. Congressional briefing. (<http://files.asme.org/asmeorg/Communities/Technical/Energy/16088.pdf>)

> Environmental Protection Agency (EPA). 2007. Greenhouse gas impacts of expanded renewable and alternative fuels use. ([http://nepis.epa.gov/Exe/ZyNET.exe/P100093S.TXT?](http://nepis.epa.gov/Exe/ZyNET.exe/P100093S.TXT?ZyActionD=ZyDocument&Client=EPA&Index=2006+Thru+2010&Docs=&Query=&Time=&EndTime=&MaximumDocuments=1&FuzzyDegree=0&ImageQuality=r75g8/r75g8/x150y150g16/i425&Display=p|f&DefSeekPage=x&SearchBack=ZyActionL&Back=ZyActionS&BackDesc=Results%20page&Maximum)

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